Edexcel iGCSE Chemistry Checklist

Section B15: **Introducing Reversible Reactions**

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| ***I can***  | ☺ | 😐 | ☹ |
| understand that some reactions are reversible and are indicated by the symbol ⇌ in equations |  |  |  |
| describe reversible reactions such as the dehydration of hydrated copper(II) sulfate and the effect of heat on ammonium chloride  |  |  |  |
| understand the concept of dynamic equilibrium  |  |  |  |
| predict the effects of changing the pressure and temperature on the equilibrium position in reversible reactions. |  |  |  |
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