Edexcel iGCSE Chemistry Checklist

Section B15: **Introducing Reversible Reactions**

|  |  |  |  |
| --- | --- | --- | --- |
| ***I can*** | ☺ | 😐 | ☹ |
| understand that some reactions are reversible and are indicated by the symbol ⇌ in equations |  |  |  |
| describe reversible reactions such as the dehydration of hydrated copper(II) sulfate and the effect of heat on ammonium chloride |  |  |  |
| understand the concept of dynamic equilibrium |  |  |  |
| predict the effects of changing the pressure and temperature on the equilibrium position in reversible reactions. |  |  |  |
|  |  |  |  |
|  |  |  |  |
|  |  |  |  |
|  |  |  |  |
|  |  |  |  |
|  |  |  |  |
|  |  |  |  |
|  |  |  |  |